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¹ Pressure-Dependent Optical and Vibrational Properties of Monolayer ² Molybdenum Disulfide

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14 **Supporting Information**



ABSTRACT: Controlling the band gap by tuning the lattice structure through pressure engineering is a relatively new route for 15 tailoring the optoelectronic properties of two-dimensional (2D) materials. Here, we investigate the electronic and lattice 16 vibrational dynamics of the distorted monolayer 1T-MoS₂ (1T') and the monolayer 2H-MoS₂ via a diamond anvil cell (DAC) 17 and density functional theory (DFT) calculations. The direct optical band gap of the monolayer 2H-MoS₂ increases by 11.7% 18 from 1.85 to 2.08 eV, which is the highest reported for a 2D transition metal dichalcogenide (TMD) material. DFT calculations 19 reveal a subsequent decrease in the band gap with eventual metallization of the monolayer 2H-MoS₂, an overall complex 20 structure-property relation due to the rich band structure of MoS₂. Remarkably, the metastable 1T'-MoS₂ metallic state remains 21 invariant with pressure, with the J_2 , A_{1g} and E_{2g} modes becoming dominant at high pressures. This substantial reversible 22 tunability of the electronic and vibrational property of the MoS₂ family can be extended to other 2D TMDs. These results 23 present an important advance toward controlling the band structure and optoelectronic properties of monolayer MoS2 via 24 pressure, which has vital implications for enhanced device applications. 2.5

26 **KEYWORDS:** 2D Materials, MoS₂, Transition Metal Dichalcogenide, Strain, Hydrostatic Pressure, Diamond Anvil Cell,

27 Pressure Engineering, Photoluminescence

A mong the great variety of exfoliable materials, transition metal dichalcogenides (TMD) have demonstrated promising optoelectronic properties. Interestingly, these properties can be largely tuned by varying the thickness,¹ by modifying the crystal structure,² or by applying strain on the TMDs.³ Pressure engineering of the two-dimensional (2D) layered crystals is still a relatively new field with many possible routes for further investigations and optimization to fully realize their potential in materials engineering.^{4–8} Being able to reversibly tune the band ₃₆ gap and structure of 2D materials by inducing pressure allows ₃₇ for exploring changes in photonic,⁹ mechanical,¹⁰ and ₃₈ electronic¹¹ properties at different hydrostatic pressures. The ₃₉

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Figure 1. Structural and vibrational properties of monolayer MoS_2 polytypes. (a) Schematic atomic arrangements of the monolayer MoS_2 showing the *a* (in-plane direction) and the cross section *c* (out-of-plane) view of the monolayer $1T-MoS_2$, $1T'-MoS_2$, and $2H-MoS_2$. (b) Raman active modes of the MoS_2 family. For the $1T'-MoS_2$, the J_1 mode is composed of two Raman active modes that are close to one another and appear as one peak at room temperature. (c) Raman active modes for the restacked monolayer $1T'-MoS_2$ and monolayer and bulk $2H-MoS_2$. The J_1 , J_2 , and J_3 modes are only prominent in the $1T'-MoS_2$ phase.

⁴⁰ effects of hydrostatic pressure on 2D materials are important to ⁴¹ understand, particularly for applications such as flexible ⁴² electronics.⁵

The transition metal dichalcogenide family, which consists of 43 44 2D layered compounds, exhibits a similar laminar structure as $_{45}$ graphene and has been investigated for several optoelec- $_{46}$ tronic^{1,12-14} and flexible^{5,15} electronics. One representative 47 TMD that has gained significant interest is molybdenum $_{48}$ disulfide (MoS₂), which is composed of a single layer of Mo 49 atoms covalently bonded between two laminar sheets of S 50 atoms. The natural bulk 2H-MoS₂ is a semiconductor that has 51 an indirect band gap of 1.3 eV, whereas its monolayer form has 52 a direct band gap of 1.85 eV due to the lack of interlayer 53 interactions in thinner layers.¹ Furthermore, depending on the 54 arrangements of sulfur atoms in the lattice, Mo can be 55 coordinated either by six sulfur atoms in either an octahedral $_{56}$ (1T), a distorted 1T (1T'), or a trigonal prismatic (2H) 57 polytype¹³ (Figure 1a). This different coordinated structure 58 also changes the electronic structure of the material. The 1T'-59 MoS₂ is metallic, whereas the 2H-MoS₂ is semiconducting.

The two main Raman modes in the monolayer $2H-MoS_2$ are 61 the A_{1g} and the E_{2g} mode. The out-of-plane A_{1g} mode occurs 62 due to opposing vibrations of the two S atoms with respect to 63 the Mo atom. The E_{2g} mode is associated with the in-plane 64 vibration of the Mo and S atoms, in opposite directions from

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one another. For the 1T'-MoS₂, the experimentally observed J₁ 65 mode involves two different types of vibrations: one at 152.68 66 cm⁻¹, which is an out-of-plane motion of each stripe of the Mo 67 atoms inside the zigzag chain, and the other mode at 157.64 68 cm⁻¹, which is an in-plane shearing vibration of one stripe of 69 atoms with respect to the other atom in the chain. Both of these 70 modes appear as a single peak at room temperature. The J₂ 71 peaks at ~226 cm⁻¹ arises from the motion of two zigzag chains 72 relative to each other, whereas the mode 330.29 cm⁻¹, named 73 J₃, tends to break each zigzag chain in two stripes with a slight 74 out-of-plane component (Figure 1b). Due to the atomic 75 arrangements of the 1T'-MoS₂ and 2H-MoS₂ phases, these 76 Raman modes with various magnitudes of activities can be used 77 to decipher the lattice dynamics of the monolayer polytypes 78 (Figure 1c).

It has been found experimentally that these two phases can 80 transition to one another via interlayer atomic plane gliding, 81 which displaces one of the S planes and changes its electronic 82 state.^{2,13,16} Another way to controllably alter the electronic 83 properties is to mechanically distort the lattice structure by 84 applying uniaxial, biaxial, or hydrostatic strain.^{17–19} The 85 majority of research in the field pertains to theoretical 86 calculations on how the distortion of the 2H and $1T'-MoS_2$ 87 lattice can lead to changes in their photonic and electronic 88 structure.^{12,19–22} Experimentally, it has been a challenge to 89



Figure 2. Pressure-dependent lattice vibrational properties of monolayer 1T'- and 2H-MoS₂. Representative Raman spectra of the monolayer (a) 2H-MoS₂ and (b) 1T'-MoS₂ at high pressures. For the 2H-MoS₂ polytype, the E_{2g} mode diminishes, whereas the dominant Raman modes are J_2 , E_{2g} and A_{1g} for 1T'-MoS₂ at pressures above 27 GPa. (c) Pressure-dependent Raman frequencies with variation in pressure for the five Raman active modes. The J_3 merges with the E_{2g} mode at around 10 GPa, whereas the J_1 diminishes at 27 GPa. The solid line indicates the theoretical agreement with the J_1 , J_2 , and J_3 modes and further supports the phonon merging of the J_3 and E_{2g} modes. (d) Applied strain is plotted with respect to the energy required to compress the in-plane and out-of-plane bonds, suggesting that the in-plane mode vibration is suppressed at higher pressures. Inset: Schematic representing the application of strain in the in-plane direction (ε_x , ε_y) (upper inset) and out-of-plane (ε_z) directions.

90 exert hydrostatic pressure homogeneously across a 2D sample, 91 and only a nominal compressive strain of 0.2% or a uniaxial 92 strain of 10% can be exerted with elaborate setups.^{10,11}

In the present study, we have investigated the MoS₂ family 93 (1T'-MoS₂ and both monolayer and bulk 2H-MoS₂) under 94 extreme hydrostatic pressures of up to 30 GPa. A diamond anvil 95 cell (DAC) with a soft neon pressure medium was used to 96 apply a hydrostatic pressure uniformly across the sample. The 97 98 DAC experiments allow fundamental studies of the effects of 99 high strain levels on the properties of materials, MoS₂ in this case. This sort of study is important for probing the intrinsic 100 response of the strain-property relations in a controlled 101 environment. The insights provided by this study can guide 102 practical realization of strained MoS₂ devices. The DAC 103 104 methodology also permits us to investigate the modulation of the band gap, the shift from a direct to an indirect band gap, 105 and the altered phononic properties of both the monolayer 2H-106 MoS₂ and 1T'-MoS₂ under applied hydrostatic pressures. 107 These results show, that for the monolayer 2H-MoS₂ polytype 108 109 a ~12% increase in band gap is observed. Although a higher pressure is not experimentally achievable for the time being, a 110 metallization at 68 GPa is predicted by theoretical calculations. 111 The semiconductor to metal transition occurs at a much higher 112 113 pressure than that in the bulk counterpart which metalizes at 114 ~19 GPa. For the $1T'-MoS_2$ polytype, the out-of-plane 115 vibrations are hindered by the hydrostatic pressure, allowing 116 only the J_2 , A_{1g} , and E_{2g} vibrational modes to become dominant

at pressures past 27 GPa. Our combined experimental and 117 theoretical results allow for a coherent understanding of the 118 vibrational, optical, and electronic properties of TMD materials 119 as a function of the number of layers and layer geometry under 120 hydrostatic pressure. 121

Raman spectroscopy, being fast and nondestructive, is a 122 powerful diagnostic tool to investigate the vibrational and 123 physical properties of MoS₂. Raman spectroscopy has been 124 used to study 2H-MoS₂ to determine the number of layers, 23 to 125 determine the effect of pressure on the 2H-MoS₂ lattice, $^{24-26}$ 126 and to determine the level of doping²⁷ and defects. 28,29 To 127 understand the electronic and mechanical stability of the MoS₂ 128 polytypes, we have studied the monolayer and bulk 2H-MoS₂ 129 as well as restacked monolayer 1T'-MoS₂ under hydrostatic 130

The lattice vibrations detected by Raman spectroscopy can $_{132}$ reveal key changes occurring to the lattice at high pressures. In $_{133}$ particular, hydrostatic pressure can induce a blue shift (phonon $_{134}$ hardening) in MoS₂.¹⁷ For the 2H-MoS₂ monolayer polytype, $_{135}$ the intensity of in-plane Raman mode (E_{2g}) starts to diminish $_{136}$ at pressures above 16 GPa (Figure 2a). The intensity ratio of $_{137}$ f2 A_{1g}/E_{2g} also increases with pressure (Supporting Information $_{138}$ Figure S1a) above 16 GPa. The diminishing intensity of the E_{2g} $_{139}$ mode is attributed to the dominance of normal compressive 140 strain on the monolayer 2H-MoS₂ under large hydrostatic 141 pressures, ³⁰ which hinders the in-plane E_{2g} movement, whereas 142



Figure 3. Pressure modulated photoluminescent properties of monolayer 2H-MoS₂. (a) Representative PL spectra as a function of pressure. An increase in the band gap with increasing pressure and then the diminishing of the PL signals at ~16 GPa indicates a direct-to-indirect band gap shift. (b) Derived band gap at high pressures. Extracted from the Lorentzian fit from the PL peaks, the band gap is shown to increase as a function of pressure. Red solid line: a polynomial fit to the experimental data described by $E_g = 1.88 + 0.03P - 8.59 \times 10^{-4}P^2$ where E_g is optical band gap and *P* is pressure. Inset: The theoretical prediction of the direct-to indirect band gap transition. (c) Band gap diagrams at K, Λ , and Γ point at two representative pressures of 0 and 22.3 GPa depicting the direct to indirect transition. Here, *D* and *I* indicate the direct and indirect band gap shift form direct (from in between K–K diagram indicates a direct-band gap between the CBM and the VBM, whereas at 22.3 GPa, band gap shift form direct (from in between K–K' points) to indirect in between K– Λ points at 22.6 GPa that agrees with the experimental observations. At around 67.9 GPa the valence band maxima (VBM) and the conduction band minima (CBM) cross the Fermi level, leading to a semiconductor-to-metal transition.

143 the A_{1g} out-of-plane vibrational mode is still prominent at 144 pressures up to 30 GPa.

Although the rate of increase is higher for A_{1g} mode 145 146 (Supporting Information Figure S1b), the E_{2g} mode also 147 increases linearly. This distinct rate of increase can be explained $_{\rm 148}$ by analyzing the type of vibrations involved in $A_{\rm 1g}$ and $E_{\rm 2g}$ 149 modes. The A_{1g} modes originate from transverse vibrations of -S atom; however, the E_{2g} mode emerges from longitudinal 150 S 151 vibrations of Mo and S atoms in opposite directions. The 152 change in Mo-S bond length (in-plane movement of atoms) plays a significant role in determining the behavior of the E_{2g} 153 154 mode. As the hydrostatic pressure increases, the out-of-plane (A_{1g}) compression becomes more favorable than the in-plane 155 (E_{2g}) compression, indicating that at higher pressures, the 156 transverse moment of S-S atom is faster than the in-plane 157 movement of Mo-S atom (Supporting Information Figure 158 S2a). This results in a higher rate of increase of A_{1g} mode with 159 160 increasing pressure in comparison to E_{2g} mode.

161 Although several studies have been done on the synthesis 162 and characterization^{13,31} of the distorted octahedral $1T'-MoS_2$ 163 polytype, the lattice vibrations at extreme hydrostatic pressure 164 conditions have not yet been studied under high pressure. 165 Previous theoretical studies have shown the breaking of lattice symmetry in 1T'-MoS₂ when tensile and compressive strain are 166 applied.²⁰ The 1T'-MoS₂ phase shows three Raman active 167 modes that are not present in the trigonal prismatic 2H-MoS₂ 168 polytype. These three extra modes are the J_1 , J_2 , and J_3 , which ¹⁶⁹ show up at 150 cm⁻¹, 225 cm⁻¹, and 325 cm⁻¹, respectively, at ¹⁷⁰ ambient pressure. As the hydrostatic pressure increases, J₃ 171 becomes indistinguishable from the E_{2g} mode at ~10 GPa, 172 whereas J_1 and J_2 are still prominent at higher pressures. The 173 broadening of the J₃ mode with pressure further supports the 174 merging phenomenon of the J_3 and E_{2g} modes (Supporting 175 Information Figure S3). In order to confirm this phonon 176 merging, theoretical calculations have been carried out to 177 investigate the effect of pressure on the vibrational properties of 178 the monolayer 1T'-MoS₂. The calculated phonon dispersion 179 curves (Supporting Information Figure S4) at ambient pressure 180 are consistent with the previously reported results³² and were 181 used to identify all the Raman active modes. The theoretically 182 derived pressure-dependent Raman frequencies show good 183 qualitative agreement with experimental observation. The 184 Raman intensity increases for the J_1 , A_{1g} , and E_{2g} modes and $_{185}$ shows the merging of the J_3 and E_{2g} modes (Supporting 186 Information Figure S5). The J₁ Raman mode also diminishes at 187 pressures above 22 GPa leaving just the J_2 , the A_{1g} and the E_{2g} 188



Figure 4. Electronic structures of the MoS_2 polytypes under hydrostatic pressure. (a) Band gap energies of the polytypes as a function of pressure. (a) Below 22.3 GPa, an increase in the direct band gap is observed for the monolayer 2H-MoS₂, whereas for higher pressure, the band gap decreases with strain and closes at 67.9 GPa indicating a metallization transition. A smaller critical pressure of 39.2 and 29.5 GPa is observed for the bilayer and trilayer 2H-MoS₂ respectively. (b) Hydrostatic pressure effects on monolayer, bilayer, trilayer, bulk MoS_2 , and $1T'-MoS_2$. A metallic state is more easily reached in the bulk state because more interlayer interactions are present.

189 Raman modes dominant at pressures past 27 GPa (Figure 2b). 190 To investigate the E_{2g} and J_3 merging in further detail, 191 theoretical calculations were performed to find the dependence 192 on pressure compared to the other modes (Figure 2c). As the 193 hydrostatic pressure increases, the out-of-plane compression 194 becomes favorable (0.08 Å/GPa) than the in-plane compression (0.03 Å/GPa), suggesting that at higher pressures the out-195 of-plane moment of the S-S atom is faster than the Mo-S 196 movement leading to a smaller rate of increase of the E_{2g} mode 197 (1.3 cm⁻¹/GPa) compared to the A_{1g} mode (2.5 cm⁻¹/GPa) 198 (Figure 2d). The pressure-dependent rate of increase for the J_1 199 200 and J₂ Raman mode is 0.9 cm⁻¹/GPa, whereas the J₃ mode 201 increases at 3.90 cm⁻¹/GPa. For the 2H-MoS₂ polytype, the 202 pressure dependence on the Raman modes of the monolayer 203 2H-MoS₂, 1T'-MoS₂ and the bulk 2H-MoS₂ shows that the A_{1g} $_{\rm 204}$ and $\rm E_{2g}$ modes for both the monolayer 2H-MoS $_{\rm 2}$ and 1T'-MoS $_{\rm 2}$ 205 polytypes are similar (Supporting Information Figure S6). For 206 the 2H-MoS₂ monolayer, the shift in Raman peak position 207 increases linearly with applied pressure. The Raman shift with 208 pressure for the A_{1g} mode is 2.6 cm⁻¹/GPa, whereas the E_{2g} 209 mode increases at a rate of 1.7 cm^{-1}/GPa (Supporting 210 Information Figure S1b). The distance between $A_{1\sigma}$ and $E_{2\sigma}$ 211 Raman modes increases at a rate of 1.2 cm⁻¹/GPa (Supporting $_{212}$ Information Figure S1c). The A_{1g} shift for monolayer MoS₂ 213 with pressure, in comparison with other 2D materials such as 214 MoSe₂, MoTe₂, and ReS₂, is higher, signifying that the monolayer MoS₂ has a strong dependence on pressure.³³ The 215 216 deviation from the bulk MoS₂ is mainly attributed to the 217 interlayer interactions along the out-of-plane axis, which is absent in the monolayer. Unlike the bulk 2H-MoS₂, no 218 intermediate phase region between the semiconducting and 219 metallic region is observed, indicating that the monolayer MoS₂ 220 does not undergo metallization or a structural transition at 221 pressures up to 30 GPa. Theoretically, we predict the 222 223 metallization of 2H-MoS₂ to be at ~68 GPa.

Due to the inherent direct band gap (1.85 eV) of the 225 monolayer 2H-MoS₂, the photonic and electronic band 226 structure can be probed by the photoluminescence (PL) 227 spectroscopy.²⁹ This strong PL emission is only observed in the 228 monolayer 2H-MoS₂ and is attributed to the slow electronic 229 relaxation that is unique to its electronic structure. The PL signal for the bilayer and trilayer are much weaker.²⁹ Our PL 230 spectral measurements on the monolayer 2H-MoS₂ show that 231 the electronic band structure can be significantly modulated by 232 hydrostatic pressure (Figure 3a) at a rate of 30 meV/GPa 233 f3 below 5 GPa. At higher pressures, the direct band gap gradually 234 increases to 2.08 eV at ~12 GPa, showing a significant 11.7% 235 change in band gap (Figure 3b), whereas increasing the PL full 236 width half-maximum (fwhm) (Supporting Information Figure 237 S1d). Although this increase in the direct band gap is reversible 238 (Supporting Information Figure S7), the rate of increase can be 239 altered depending on the pressure medium. This can 240 significantly modify the photonic and electronic characteristics ²⁴¹ of 2D materials under high pressure.^{6,9,34} Because we have used ²⁴² an inert, soft Ne pressure medium in our experiments, our 243 results therefore represent more intrinsic features of the system 244 without dopant effects. The polynomial increase in the band 245 energy is in agreement with theoretical DFT calculations, which 246 show an increase in the direct band gap at the K-K' point. The 247 theoretical band gap is underestimated because the presence of 248 artificial self-interaction and the absence of the derivative 249 discontinuity in the exchange-correlation potential with the 250 PBE/GGA methods.³⁵ The initial trend in band gap as 251 supported by the PL experiments (Figure 3b) shows an 252 increase at the K-K' point. The in situ fwhm map of PL 253 Spectra (Supporting Information Figure S8) increases from 254 24.7 nm at 0.6 GPa to 58.7 nm at 12.5 GPa and diminishes to 255 the background noise level at pressures over 16 GPa at room 256 temperature. This increase in the fwhm implies a crossover 257 point to the direct-to-indirect band gap³ at the K-K' point at 258 approximately 16 GPa. This direct to indirect band gap shift 259 was further conformed by DFT calculations. The ambient 260 theoretical band gap initially lies in-between K-K' point until a 261 pressure of 22.3 GPa is reached. At 22.3 GPa, a direct-to- 262 indirect transition indicates that the conduction band minima 263 (CBM) and the valence band maxima (VBM) are no longer at 264 the K–K' interband (Figure 3c), but rather shift to the K–A $_{265}$ point. Theoretical calculations reveal that the band gap of 2H- 266 MoS_2 monolayer first increases up to a pressure of ~22 GPa is 267 reached. Beyond this pressure, the band gap gradually 268 decreases, eventually leading to the metallization of monolayer 269 2H-MoS₂ at around 68 GPa. 270

271 To understand the mechanism for this band gap change and 272 the semiconductor to metal (S-M) transition under hydro-273 static pressure, we analyze the contribution from different 274 molecular orbitals by performing angular momentum projected 275 density of states (LDOS) calculations. At lower pressures, both valence (VB) and conduction bands (CB) are mainly composed of Mo- $d_x^2 - y^2$, d_z^2 and S- p_x , p_y orbitals (Figure 3c). 277 278 With increasing pressure, these orbitals move away from the 279 Fermi level, resulting in an increase in band gap up to 22.3 GPa pressure. At higher pressure, the out-of-plane compression 280 becomes more dominant (Supporting Information Figure S2b), 281 282 which makes the Mo d_z^2 , d_{xz} , and p_z orbitals interact strongly with S p_z and p_x orbitals. This strong intralayer hybridization 283 284 between the orbitals mentioned above leads to a decrease in 285 band gap after ~22 GPa pressure and finally closing of the band gap at higher pressure (68 GPa). For the monolayer $2H-MoS_{2}$, 286 287 in-plane intra layer interactions play a dominant role in the 288 metallization at 67.9 GPa.

It was found in our previous study of the bulk MoS₂ that 289 interlayer interactions play a critical role in the metallization.¹⁷ 290 291 It is therefore expected that as the number of layers increases, 292 the critical pressure or critical volume strain (Supporting Information Figure S10) at which the metallization occurs 293 decreases (Figure 4a). The metallization of the bilayer, trilayer, 294 and bulk 2H-MoS₂ (Figure 4a and b) under high pressure can 295 be understood by the interaction of electron-donating sulfur 296 atoms between the van der Waals gaps.^{17,21} The transition 297 298 pressure for MoS₂ therefore decreases as the number of layers 299 increases. As experimentally observed, the monolayer 1T'-300 MoS₂ shows metallic behavior (Supporting Information Figure 301 S7), unlike the 2H-MoS₂ polytype which is semiconducting in 302 nature. For the 2H-MoS₂, the valence and conduction bands $_{303}$ are mainly composed of the *d* orbitals of Mo and *p* orbitals of S 304 atoms (Supporting Information Figure S11a). Upon application 305 of hydrostatic pressure the extent of hybridization between Mo 306 *d* orbitals (mainly d_z^2 , d_{xy} , and $d_x^2 d_{x-y}^2$) and S *p* orbitals increases, 307 thus increasing the overlap between conduction and valence 308 bands resulting in an enhanced metallization (Supporting 309 Information Figure S11b).

We have measured an 11.7% difference in the direct band 310 311 gap of the 2H-MoS₂ from 1.85 eV at ambient pressure to 2.08 312 eV at 16 GPa before the PL signal diminished and could no 313 longer be resolved at higher pressures and room temperature. 314 For comparison, a 3.3% difference in direct band gap was found 315 by S.P. Lau et al.,¹¹ a 3.0% difference by G. A. Steele et al.,³⁶ 316 and a 5.4% difference by Dou et al.⁹ for the monolayer 2H-317 MoS₂. To further understand the electronic and phononic 318 properties, DFT calculations were conducted to show that a 319 critical pressure of 67.9 GPa is required to metalize the 320 monolayer 2H-MoS₂. The 1T'-MoS₂ behaves as a metal at 321 ambient pressure and exhibits three more Raman active modes 322 $(J_1, J_2, and J_3)$ in comparison to the 2H-MoS₂, which exhibits $_{323}$ just the A_{1g} and E_{2g} modes. At pressures above 10 and 27 GPa, $_{324}$ we have observed the diminishing of the out-of-plane J_1 and J_3 325 Raman modes, respectively, due to higher susceptibility to 326 hydrostatic pressure along the out-of-plane axis. At 10 GPa, the $_{327}$ J₃ mode starts to merge with the E_{2g} mode. The suppression of 328 the J_1 mode at ~27 GPa is attributed to the large hydrostatic 329 pressure on the lattice implying that the effect of pressure is 330 more pronounced for this mode and is more sensitive to 331 pressure. Because this process is reversible for both the 2H- $_{332}$ MoS₂ and the 1T'-MoS₂, the distortion of the lattice is 333 predicted to be a transitory structural modification. The total

energy calculation also reveals that monolayer MoS₂ is the most 334 stable phase under the applied pressure range (Supporting 335 Information Figure S12), and confirms that there is no phase 336 transformation between 2H and 1T' phase within the studied 337 range of pressure. 338

Strain engineering of 2D layered materials is a relatively new 339 field with possible routes for further investigations to fully 340 realize their potential in customizing material behavior. Here, 341 the interplay of the structural and optoelectronic properties is 342 successfully described for the MoS₂ family. The monolayer 2H- 343 MoS₂ differs from its bulk counterpart because the bandgap 344 increases by $\sim 12\%$ with applied hydrostatic pressure due to the 345 absence of interlayer interactions. For the intrinsically metallic 346 1T'-MoS₂, interlayer structural distortion is prevalent resulting 347 in phonon merging of two Raman active modes (J $_1$ and E $_{2g}$) at $_{348}$ high pressures. The large reversible bandgap modulation of 349 semiconducting MoS₂ is attractive for highly tunable optoelec- 350 tronics, nanoelectronics, and flexible electronics and can enable 351 new opportunities based on the dynamic strong electron- 352 phonon coupling. 353

Experimental Methods. Sample Preparation. The 2H- $_{354}$ MoS₂ monolayer were synthesized based on our previous $_{355}$ work.³⁷ In brief, 300 nm SiO₂ substrates were placed in the $_{356}$ center of a tubular furnace. Precursors of 0.3 g MoO₃ (Sigma- $_{357}$ Aldrich, 99.5%) in Al₂O₃ crucible was placed 12 cm away from $_{358}$ substrates and S (Sigma-Aldrich, 99.5%) powder in quartz tube $_{359}$ was evaporated in tube and reacted with Al₂O₃. The furnace $_{360}$ was first heated to 150 °C at 10 °C/min rate with 70 sccm Ar at $_{361}$ 10 Torr and annealed for 20 min, then reached 650 °C at 25 $_{362}$ °C/min rate and kept for 1 h. After growth, furnace was slowly $_{363}$ cooled to room temperature. The as-grown MoS₂ film was $_{364}$ broken into thin pieces that were used after cleaning the $_{365}$ substrate with acetone. The thinned down sample was $_{366}$ transferred over to the DAC by a 10 μ m edged tweezer and $_{367}$ was studied as is.

1T'-MoS₂ samples were synthesized by Li intercalation using 369 *n*-butyllithium as explained elsewhere.¹³ The Li_xMoS₂ was 370 retrieved by filtration and washed with hexane (60 mL) to 371 remove excess *n*-butyllithium and other organic residues. 372 Exfoliation was achieved by ultrasonicating Li_xMoS₂ in water 373 at a concentration of 1 mg/mL for 1 h. The mixture was 374 centrifuged several times to remove lithium cations in the form 375 of LiOH as well as nonexfoliated materials. 376

Characterization. For the Raman spectroscopic studies on 377 the 2H and 1T' polytypes, a green 532 nm Coherent Verdi V2 378 laser was used at the Mineral Physics Laboratory of the 379 University of Texas at Austin. The scattered light was dispersed 380 by an 1800 grooves/mm grating and collected by an EMCCD 381 (Andor Technology) resulting in a spectral resolution of 382 approximately 1 cm⁻¹. To avoid potential overheating or 383 oxidizing of the samples, the highest laser power level used was 384 20 mW with a focused laser beam size of 10 μ m. Ruby 385 fluorescence spectra were also collected using the same system 386 for pressure calibrations³⁸ (Supporting Information Figure S8). 387

For the photoluminescence (PL) and Raman in situ mapping 388 studies on the 2H-MoS₂, a grating of 900 grooves/mm was 389 used. A long working-distance (LD EC) objective, Epiplan- 390 Neofluar (20x/0.22 DIC M27), was used. The Witec Alpha 300 391 micro-Raman confocal microscope with a 488 nm laser was 392 used for the in situ PL and Raman mapping studies. To better 393 understand the band gap energy of the system, a Lorentzian 394 curve fit to the PL curves was used to derive the peak energy 395 positions as well as the band widths.³⁹

Theoretical Calculations. Modeling of Hydrostatic 397 398 Pressure on Monolayer MoS₂. The hydrostatic pressure on the monolayer was modeled by calculating the ease of 399 compression along the x, y, and z direction. First, we checked 400 the individual energy costs for in-plane (x, y) and out of plane 401 compression (z). Based on the energy cost calculation we 402 calculated the compression along all directions for hydrostatic 403 pressure modeling (Supporting Information Figure S2a and 404 S2b). A similar procedure was followed for 1T'- MoS₂ 405 406 monolayer.

For 2, 3L, and bulk MoS₂, first we approximate the in-plane 407 (change in a, b) and out-of-plane (change in slab thickness c) 408 compression by performing total energy calculation. First, we 409 410 apply a uniform strain in all direction calculate the energy cost. 411 By comparing the energy cost for biaxial and normal compressive strain, we extract the atomic displacement. Due 412 to presence of weak van der Waals (vdW) interaction in 413 between the layers the slab thickens decreases more rapidly in 414 comparison to in plane lattice parameter a and b as shown in 415 416 Supporting Information Figure S13 (a) for 2L MoS₂.

⁴¹⁷ In order to correlate the theoretical finding with experiments, ⁴¹⁸ it is important to estimate applied hydrostatic pressure. The ⁴¹⁹ applied pressure was calculated from the energy cost per unit ⁴²⁰ volume as per the following equation for 1L, 2L, 3L, and bulk ⁴²¹ MoS₂

$$P = \frac{E - E_0}{V - V_0}$$

⁴²² where, $E(E_0)$ and $V(V_0)$ are the energy and volume for the ⁴²³ unstrained (strained) systems. The volume of a hexagonal ⁴²⁴ lattice can be defined as $V = a^2 c \sin(60)$ where *a* is the in-plane ⁴²⁵ lattice parameter and *c* is thickness of slab. The thickness of ⁴²⁶ various structures is defined as shown in Supporting ⁴²⁷ Information Figure S13. As shown in our previous work,¹⁷ ⁴²⁸ the theoretical calculated pressure agrees well with the ⁴²⁹ experimental pressure for bulk MoS₂.

First-Principles Theoretical Calculations. Theoretical calcu-430 431 lations were performed using first-principle ab initio density 432 functional theory (DFT) using Vienna Ab Initio Simulation 433 Package (VASP).⁴⁰ All-electron projector augmented wave 434 potentials^{41,42} and the Perdew-Burke-Ernzerhof⁴³ generalized 435 gradient approximation (GGA) were used to account for the 436 electronic exchange and correlation. The structure of the 1T'-437 MoS₂ monolayer is generated by following the approach of Calandra.²³ All the structures were relaxed in the x-y plane 438 while constraining the z coordinates by employing a conjugate 439 gradient scheme until the forces on every atom were minimized 440 441 to be less than 0.005 eV/Å. A well-converged Monkhorst-Pack $_{442}$ k point set of $15 \times 15 \times 1$ was used for this procedure. The 443 phonon dispersion of monolayer MoS₂ was calculated using 444 density functional perturbation theory (DFPT)⁴⁴ as imple-445 mented in VASP. An additional tool, Phonopy,45 which 446 supports the VASP interface, was used for extracting the 447 phonon frequencies. The force constants were calculated with a 448 supercell of $2 \times 2 \times 1$ and the k point set of $13 \times 13 \times 1$. In 449 order to obtain accurate phonon frequencies, a high energy 450 cutoff of 600 eV and strict energy convergence criterion of $451 \ 10^{-11} \text{ eV}$ were used.

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ASSOCIATED CONTENT 452 Supporting Information 453

Here, you will find further experimental and theoretical support 454 for the optical and structural properties discussed in this 455 manuscript. This material is available free of charge via the 456 Internet at http://pubs.acs.org. 457

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