Site-Specific Pressure-Driven Spin-Crossover in Lu_{1-x}Sc_xFeO₃

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ABSTRACT: Pressure-driven spin-crossover (PSCO) is a collective quantum phenomenon frequently observed in transition-metal-based systems. According to the crystal-field theory, PSCO highly depends on the surrounding coordination environment of a given magnetic ion; nevertheless, it has never been verified experimentally up to now. Herein, we report the observation of a site-specific PSCO phenomenon in $Lu_{1-x}Sc_xFeO_3$, in which octahedrally coordinated Fe^{3+} in orthorhombic $LuFeO_3$ and trigonal-bipyramidally coordinated Fe^{3+} in hexagonal $Lu_{0.5}Sc_{0.5}FeO_3$ show distinct PSCO response to external pressure. X-ray emission spectra and DFT calculations reveal the key role of coordination environment in a PSCO process and predict the occurrence of PSCO for trigonal-bipyramidally coordinated Fe^{3+} in LuFeO₃. The demonstration of site-specific PSCO sheds light on the state-of-the-art design of PSCO materials for directional applications.



S pin-crossover (SCO), as a consequence of the d-orbital splitting in a ligand field, is generally characterized by transition-metal centers undergoing a spin state transition in response to external stimuli such as heat, light, pressure, and magnetic fields.¹⁻⁴ Typically, SCO is observed in first-row transition-metal complexes with electronic configurations d⁴ to d⁷. To date, the well-studied SCO system is Fe²⁺(d⁶)-based complexes with an FeL₆ coordination manner.^{5,6} Because the molecular bistable status of SCO is associated with a lack of fatigue, SCO materials hold great promise for applications in sensors, displays, and digital memory. Particularly, when a classical light- or temperature-initiated SCO arises along with structural phase transitions and/or physical property alterations, SCO materials can become multifunctional.⁷⁻¹⁰

Although not as common as light- or temperature-initiated SCO, pressure-driven SCO (PSCO) has also been frequently reported in many material systems. Pressure, as an alternative external stimulus to light or temperature, is the most direct route to altering the crystal-field strength by shortening the metal-ligand distances and thus can drive SCO especially in strongly correlated systems.^{11,12} That is, benefitting from the power of pressure, the exploration of PSCO material has no longer been limited to the soft metal-organic complex system. Surprisingly, to date, PSCO has been reported more as only a physical phenomenon¹³⁻¹⁹ rather than a useful functionality for modern material design. We have focused on this issue for a long time and successfully developed a series of materials with abrupt PSCO following the principle to reduce the dimensionality of metal centers to increase the "cooperativity" between them.^{20–22}

According to the crystal-field theory, the ground spin state of a metal center embedded in a crystalline matrix is determined by the relative magnitude of the d-orbital splitting energy (Δ) along with the Hund's intra-atomic exchange energy (J).^{23,24} In weak fields ($\Delta < J$), the ground state is high spin (HS) with a maximum spin multiplicity, whereas strong fields $(\Delta > I)$ stabilize the low spin (LS) state with a minimum multiplicity. For different d-orbital splitting situations, the effect of pressure on the compression deformation behavior of metal-ligand polyhedra and also the change of Δ is different. Therefore, PSCO should be highly dependent on the surrounding coordination environment of a given magnetic ion; nevertheless, this viewpoint has never been experimentally verified to date. In this work, we studied the site-specific PSCO phenomenon in $Lu_{1-x}Sc_xFeO_3$, which perfectly provides two distinct coordination environments for Fe^{3+} (octahedral at x =0 and trigonal-bipyramidal at x = 0.5, respectively). In situ synchrotron X-ray diffraction (XRD) and X-ray emission spectroscopy (XES) experiments were performed to verify the PSCO behavior of Fe³⁺. Density function theory (DFT) calculations were also conducted to reveal the spin state evolutions of Fe³⁺ in different crystal fields under high pressure.

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 $Lu_{1-x}Sc_xFeO_3$ possesses a characteristic structure sequence of rare-earth orthoferrites along with the increase of Sc content.^{25–30} LuFeO₃ and Lu_{0.5}Sc_{0.5}FeO₃ fine powders were synthesized via a sol-gel method followed by high-temperature sintering (for details, see the Supporting Information). XRD analyses prove their high phase purity and crystallinity in orthorhombic and hexagonal symmetry, respectively (Figure S1). The crystal structures of LuFeO₃ and Lu_{0.5}Sc_{0.5}FeO₃ are shown in Figure 1. LuFeO₃ adopts a distorted perovskite



Figure 1. Crystal structures of (a) orthorhombic LuFeO₃ with octahedrally coordinated Fe³⁺ and (b) hexagonal Lu_{0.5}Sc_{0.5}FeO₃ with trigonal-bipyramidally coordinated Fe³⁺.

structure in which all of the Fe³⁺ ions coordinate with six O²⁻ anions in an undistorted octahedral geometry. Comparatively, the hexagonal Lu_{0.5}Sc_{0.5}FeO₃ comprises closely packed layers of vertex-sharing FeO₅ bipyramids to form triangular lattices in the *ab* plane, which are separated by wrinkled Lu/Sc layers along the *c*-axis. All the Fe³⁺ ions are surrounded by three inplane and two apical O²⁻ ligands. LuFeO₃ and Lu_{0.5}Sc_{0.5}FeO₃ provide a rare example of two analogues similar in composition but distinct in local structure, which enables the comparative study of site-specific PSCO.

In situ XES measurements on both samples loaded inside diamond anvil cells (DACs) were performed to probe the spin state evolution under high pressure and room temperature (for details, see the Supporting Information). Figure 2 presents the Fe $K\beta$ XES spectra of LuFeO₃ and Lu_{0.5}Sc_{0.5}FeO₃ under compression up to 60 GPa, respectively. Typically, the XES spectra at low pressures comprise a main $K\beta_{1,3}$ peak located around 7058.6 eV and a satellite $K\beta'$ peak located around 7044.6 eV. It is well-known that the $K\beta$ lines are characteristic



Figure 2. Fe K_{β} XES of (a) LuFeO₃ and (b) Lu_{0.5}Sc_{0.5}FeO₃ as a function of applied pressure. Inset of panel a shows the mechanism of an XES measurement to reflect the spin states of a 3d metal ion.

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X-ray emissions originating from the 3p to 1s decay.⁴ The parameters of $K\beta$ lines, especially the intensity of $K\beta'$ peaks, can reflect the interactions between the 3p core hole and the partially filled 3d shell electrons (inset of Figure 2). Therefore, the occurrence of an HS-to-LS transition can be distinguished simply by the decrease of $K\beta'$ peaks and the left-shift of $K\beta_{1,3}$ peaks. For LuFeO₃, well-defined $K\beta'$ satellite peaks are observed in the low-pressure range (below 44 GPa), indicating a HS ground state (S = 5/2). When pressure is greater than 48 GPa, the Fe K β' peak starts to decrease and becomes indistinguishable above 52 GPa, and the Fe K $\beta_{1,3}$ lines shift to lower energies correspondingly. This evidence indicates the complete occurrence of PSCO of Fe^{3+} (S = 5/2 to S = 1/2) in LuFeO3 around 50 GPa. Whereas, for Lu05Sc05FeO3, no decrease of the Fe K β' peak or shift of the Fe K $\beta_{1,3}$ peak is observed up to 60 GPa, indicating the HS state of Fe³⁺ (S = 5/ 2) in hexagonal $Lu_{0.5}Sc_{0.5}FeO_3$ can survive to a higher pressure than that in orthorhombic LuFeO₃. The absence of PSCO in Lu_{0.5}Sc_{0.5}FeO₃ reveals the key impact of coordination environment surrounding transition-metal ions on the onset of a PSCO incident.

Structure analyses based on in situ XRD patterns provide a mutual corroboration along with XES evidence on the occurrence of PSCO in LuFeO3 and the absence of PSCO in Lu_{0.5}Sc_{0.5}FeO₃. LuFeO₃ suffers an iso-structural phase transition around 49 GPa as evidenced by the abrupt shifts of representative XRD peaks (110), (002), and (111) to higher 2θ angles, corresponding to collapses of the cell parameters and the cell volume (Figures S2 and S3). This is in accordance with many previous experiences that PSCO usually occurs along with a lattice collapse.³¹ Although there is no symmetry change, octahedral tilts of FeO₆ are very possible in the isostructural phase transition referring to previous studies.³² Lu_{0.5}Sc_{0.5}FeO₃ also undergoes an iso-structural phase transition around 45 GPa accompanied by a volume collapse (Figures S4 and S5). However, this volume collapse is not induced by the moment collapse of Fe³⁺, but instead a distortion of FeO₅ trigonal bipyramids with elongation of the apical Fe-O bonds along the *c*-axis.

Quantitative analyses of the XES data were conducted following the IAD procedure described by Vankó et al.,³³ by which the IAD value reflects a linear relationship with the average spin quantum number, S. Figure 3a shows the pressure dependence of the Fe^{3+} spin state in LuFeO₃ and



Figure 3. (a) Spin state fractions and IAD values derived from the Fe K_{β} XES spectra of LuFeO₃ and Lu_{0.5}Sc_{0.5}FeO₃ as a function of pressure. (b and c) Schematic diagrams of the d orbital splitting of Fe³⁺ in octahedrally and trigonal-bipyramidally coordinated environments, respectively.

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Lu_{0.5}Sc_{0.5}FeO₃ from the IAD analysis. At low pressures, Fe³⁺ ions in both materials exhibit a high spin state with S = 5/2(IAD \approx 0). As the pressure approaches 50 GPa, the IAD value for LuFeO₃ increases abruptly indicating the occurrence of PSCO for Fe^{3+} in the octahedral sites. For Lu_{0.5}Sc_{0.5}FeO₃, the IAD value remains close to 0 and the high spin fraction remains nearly 100% up to more than 60 GPa. The site-specific PSCO in LuFeO₃ and Lu_{0.5}Sc_{0.5}FeO₃ can be explained by the crystal-field theory. It is well-known that the spin state of a transition-metal ion depends on the competition between the d-orbital splitting energy ($\Delta = 10$ Dq) and the electronic Coulomb repulsion (J). In an octahedral coordination field for Fe^{3+} ions in LuFeO₃, the five d orbitals split into 3-fold degenerated d_{xy} , d_{yz} , and d_{xz} and 2-fold degenerated $d_{x^2-y^2}$ and d_{z^2} (Figure 3b). In this configuration, high-pressure drives the increase of Δ markedly and most iron(III) oxides with an octahedral coordination manner undergo PSCO between 35 and 50 GPa. In the rare trigonal bipyramidal coordination field for Fe^{3+} ions in hexagonal $Lu_{0.5}Sc_{0.5}FeO_3$, the five d orbitals split into two lower d_{xz} and d_{yz} , two intermediate d_{xy} and $d_{x^2-y^2}$, and a highest d_{z^2} (Figure 3c). Although Δ is always a little higher than that within an octahadral field, trigonalbipyramidally coordinated Fe³⁺ ions still adopt HS state at ambient conditions. It is worth mentioning that, under high pressure, one electron may jump from d_{z^2} to the lowest d_{xz} or d_{yz} , which results in an intermediate spin (IS) state with S = 3/2 rather than a LS state with S = 1/2. During the LP-to-HP phase transition (very likely HS-to-IS), the elongation of the apical Fe-O bonds within FeO₅ trigonal bipyramids leads to lower d_{z^2} orbital and thus decrease Δ (Figure S6), which inhibits the occurrence of PSCO in hexagonal Lu_{0.5}Sc_{0.5}FeO₃. After that, high pressure will keep shortening all the Fe-O bonds (increasing Δ) and finally leads to the occurrence of PSCO (HS-to-IS) at a higher pressure. Moreover, we mention the possible impact of deviatoric stress, frequently observed in nanosystems,^{34,35} on the occurrence of structural phase transition and PSCO in Lu_{1-x}Sc_xFeO₃. The use of silicone oil as the pressure-transmitting medium (PTM) may slightly affect the occurrence pressure of the iso-structural phase transitions and the steepness of PSCO in $Lu_{1-r}Sc_rFeO_{3r}$ whereas the phenomenon of site-specific PSCO will not be changed by the PTM effect.

DFT calculations based on the HS, IS, and LS states of Fe³⁺ were performed to validate the role of coordination environments in the PSCO process of Fe3+ within LuFeO3 and Lu_{0.5}Sc_{0.5}FeO₃. For simplicity and ease of comparison, two LuFeO₃ structures, one with an orthorhombic symmetry (ICSD# 27285)³⁶ and the other with the hexagonal Lu_{0.5}Sc_{0.5}FeO₃ structure were adopted for the DFT calculations (for details, see the Supporting Information). LDA+U method with U = 4.3 eV was adopted by comparing the experimental cell volumes with those from calculations (Figure S7). For orthorhombic LuFeO₃ with octahedrally coordinated Fe³⁺, a pressure-driven HS-to-LS transition occurs around 60 GPa, meeting but a little higher than the experimental result (Figure 4a). For hexagonal LuFeO₃ with trigonal-bipyramidally coordinated Fe³⁺, first, the HS state remains in the ground state up to 70 GPa within the experimentally verified pressure range. Linear extension predicts a PSCO point (HS-to-IS transition) of Fe³⁺ around 100 GPa, and finally to a LS state at very high pressure (more than 150 GPa) (Figure 4b). Both our experimental and calculation results reveal the site-specific feature of the PSCO phenomenon. It is expected that a



Figure 4. Relative total energy calculated for the HS, IS, and LS states in (a) orthorhombic LuFeO₃ with octahedrally coordinated Fe^{3+} and (b) hexagonal LuFeO₃ with trigonal-bipyramidally coordinated Fe^{3+} .

transition-metal ion within different crystal field will exhibit distinct PSCO behaviors, whereas it is unexpected that there would be such a big difference between the two situations (60 GPa in octahedron versus 145 GPa in trigonal-bipyramid). Benefiting from the difference brought by crystal field strength, it is possible to design tailored PSCO materials for specific application scenarios.

In summary, we report a site-specific PSCO phenomenon in the $Lu_{1-x}Sc_xFeO_3$ system for the first time. Besides the similar chemical composition, LuFeO3 and Lu05Sc05FeO3 have distinct coordination environments for Fe³⁺ ions: octahedrally coordinated in orthorhombic LuFeO3 and trigonal-bipyramidally coordinated in hexagonal Lu_{0.5}Sc_{0.5}FeO₃. They show distinct PSCO as evidenced by both experimental results (XES and XRD) and theoretical calculations. Fe³⁺ within the octahedral FeO₆ sites undergoes a PSCO around 50 GPa (S = 5/2 to S = 1/2). Comparatively, no PSCO occurs experimentally up to 60 GPa for Fe³⁺ within the trigonalbipyramidal FeO₅ sites, which is predicated to happen above 100 GPa. These results validate the key role of coordination environment in a PSCO process. The demonstration of sitespecific PSCO paves the way to precise design of novel pressure-responsive switch materials for various practical applications.

EXPERIMENTAL METHODS

Sample Preparation. $LuFeO_3$ and $Lu_{0.5}Sc_{0.5}FeO_3$ powders were synthesized using a sol-gel method followed by hightemperature sintering. The phase purity was confirmed by powder XRD at room temperature and ambient pressure.

High-Pressure Experiments. A symmetrical diamond anvil cell (DAC) was employed to generate high pressure. A steel gasket was preindented to about 40 μ m in thickness followed by laserdrilling the central part to form a 120 μ m diameter hole to serve as the sample chamber. Precompressed LuFeO₃ and Lu_{0.5}Sc_{0.5}FeO₃ pellets and ruby ball were loaded in the chamber. Silicone oil was used as the pressure-transmitting medium. The *in situ* angle-dispersive X-ray diffraction (ADXRD) experiments were carried out at the BL15U1 beamline station of Shanghai Synchrotron Radiation Facility (SSRF) at room temperature. A focused monochromatic X-ray beam with wavelength of 0.6199 Å was used for the diffraction

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experiments, and the ADXRD patterns were recorded by a MAR165 CCD detector. For the high-pressure X-ray emission spectrum (XES) experiments performed at 16 ID-D station of HPCAT, Be gaskets were used as the sample chamber and silicone oil was used as pressure-transmitting medium.

DFT Calculations. Two LuFeO₃ structures, one with an orthorhombic symmetry and the other with the hexagonal Lu_{0.5}Sc_{0.5}FeO₃ structure were adopted for the DFT calculations. DFT calculations were performed using the VASP package, and the electron–electron interaction is treated using the LDA+U with U = 4.3 eV; J = 0; energy cutoff, 600 eV; and k points, $6 \times 6 \times 4$. The PAW–PBE pseudopotential is used, and valence electronics of Lu, Fe, and O are Sp⁶Sd¹6s², 3d⁷4s¹, and 2s²2p⁴, respectively.

ASSOCIATED CONTENT

1 Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jpclett.0c02537.

Experimental details, characterization data, theoretical calculations, and supplemental figures (PDF)

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Notes

The authors declare no competing financial interest.

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